

TBAT	Outcomes	Skills	Assessment
1. Explain the differences between atoms, elements, compounds and mixtures.	<p>State that all matter is made up of tiny particles called atoms.</p> <p>Explain the differences between elements, compounds and mixtures (with reference to elements being substances that cannot be broken down into anything simpler by chemical means).</p> <p>Represent atoms, molecules of elements and simple compounds using a model.</p>		Self assessment.
2. Describe how elements are represented in the periodic table.	<p>Recall that elements are often represented by symbols.</p> <p>Use the periodic table to look up symbols for elements.</p> <p>Explain how new evidence has changed ideas about elements.</p>	Using the periodic table.	Self Assessment
3. Construct word equations.	<p>Identify the products and reactants using a word equation.</p> <p>Supply missing reactants or products to complete a word equation.</p> <p>Model simple reactions using word equations.</p>	Constructing chemical equations.	Self Assessment
4. Investigate the properties of elements, mixtures and compounds.	Describe how elements can combine to form compounds.	Working scientifically: Iron + Sulfur -> Iron sulphide investigation	Teacher assessed

	<p>Describe the changes in properties between a compound and its constituent elements.</p> <p>Apply knowledge of thermal decomposition in carbonates to other compounds.</p>	6 mark q: Explain the difference in properties between elements, compounds and mixtures.	
5. Therapy	Personalised reflection based upon the teacher assessed piece.		
6. Investigate chemical and physical changes.	<p>Recall examples of chemical reactions in everyday life.</p> <p>Recall some observations that indicate a chemical reaction.</p> <p>Explain the difference between chemical reactions and physical changes.</p>	Working scientifically	Self Assessed.
7. Explain the process of thermal decomposition.	<p>Identify thermal decomposition reactions.</p> <p>Describe what happens during thermal decomposition of a metal carbonate.</p> <p>Use a symbol equation to model the reactions.</p>	Working scientifically.	Self Assessed.
8. Describe the properties of metals and non-metals.	<p>Describe some common properties of metallic and non-metallic materials.</p> <p>Use ideas about the periodic table to identify the positions of metal and non-metal elements.</p> <p>Relate the uses of different elements to their properties</p>		Self assessed
9. Investigate metals and non-metals.	Describe the evidence needed to decide whether an element is a metal or a non-metal.	Working scientifically.	Self Assessed

	<p>Use evidence to classify unfamiliar materials as being metal elements, metallic, non-metal elements, non-metallic.</p> <p>Interpret experimental evidence to identify elements.</p>		
10. Revise atoms, elements and compounds.	<p>Complete the revision mat to consolidate learning.</p> <p>Complete the quick quiz to assess their understanding and focus their revision</p>		Self assess
11. Assessment	End of unit test	Exam technique	Peer assessed
12. Reflection lesson	Students to reflect on their assessment using the knowledge organiser		Self assess